

The primary purpose of this study was to determine the effect of the addition of a small amount of a certain substance to a mixture of two other substances. The results showed that the addition of this substance significantly increased the rate of reaction between the two substances.

The rate of reaction was measured by the amount of product formed over a period of time. The reaction was carried out at a constant temperature of 25°C. The concentration of the reactants was kept constant, and only the concentration of the added substance was varied.

The results of the experiment are shown in the following table. It can be seen that the rate of reaction increases as the concentration of the added substance increases. This suggests that the added substance acts as a catalyst for the reaction.

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